

5.3 – Simple Gas Laws II - Charles' Law - Worksheet

- 1) The temperature inside my refrigerator is about 4°C . If I place a balloon in my fridge that initially has a temperature of 22°C and a volume of 0.5 liters, what will be the volume of the balloon when it is fully cooled by my refrigerator?

- 2) A man heats a balloon in the oven. If the balloon initially has a volume of 0.4 liters and a temperature of 20°C , what will the volume of the balloon be after he heats it to a temperature of 250°C ?

- 3) On hot days, you may have noticed that potato chip bags seem to “inflate”, even though they have not been opened. If I have a 250 mL bag at a temperature of 19°C , and I leave it in my car which has a temperature of 60°C , what will the new volume of the bag be?

- 4) A soda bottle is flexible enough that the volume of the bottle can change even without opening it. If you have an empty soda bottle (volume of 2 L) at room temperature (25°C), what will the new volume be if you put it in your freezer (-4°C)?

- 5) Some students believe that teachers are full of hot air. If I inhale 2.2 liters of gas at a temperature of 18°C and it heats to a temperature of 38°C in my lungs, what is the new volume of the gas?
- 6) How hot will a 2.3 L balloon have to get to expand to a volume of 400 L? Assume that the initial temperature of the balloon is 25°C .
- 7) I have made a thermometer which measures temperature by the compressing and expanding of gas in a piston. I have measured that at 100°C the volume of the piston is 20 L. What is the temperature outside if the piston has a volume of 15 L? What would be appropriate clothing for the weather?
- 8) To what temperature must 50.5 mL of N_2 gas at 33.4°C and 1.00 atm pressure be heated to produce the same percentage change in volume as observed when 35.0 mL of O_2 gas at 1.00 atm pressure is heated from -13.5°C to 25.0°C .